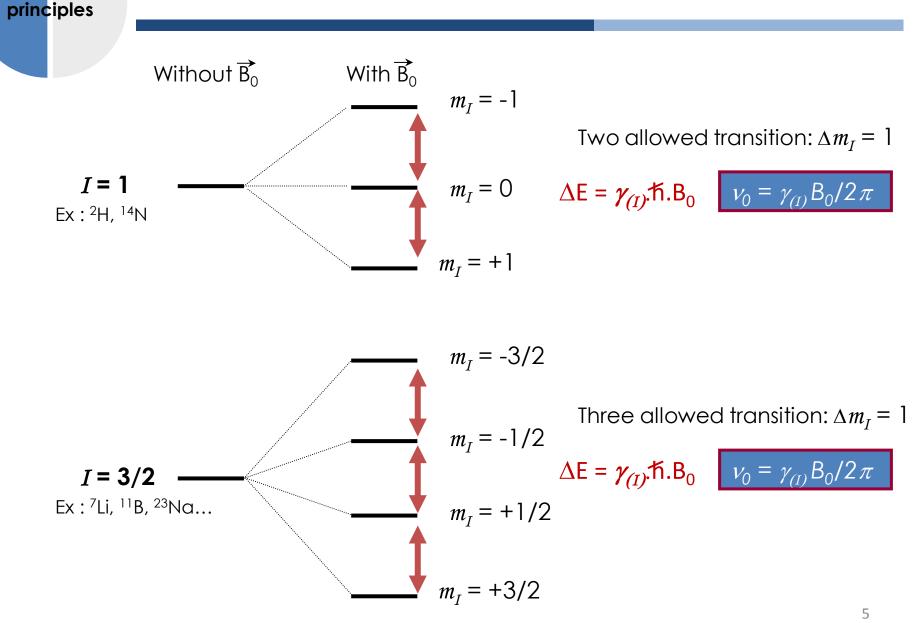
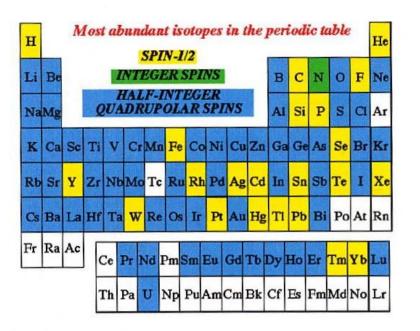
2. Zeeman effect

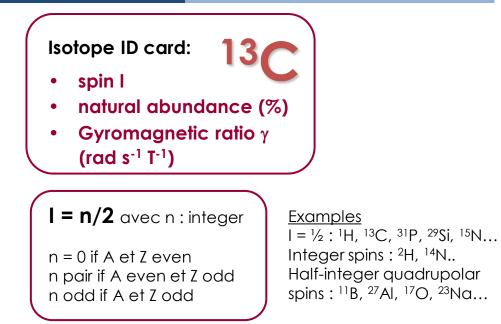
NMR

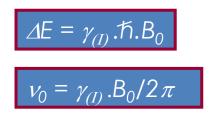


NMR principles

3. Isotopes





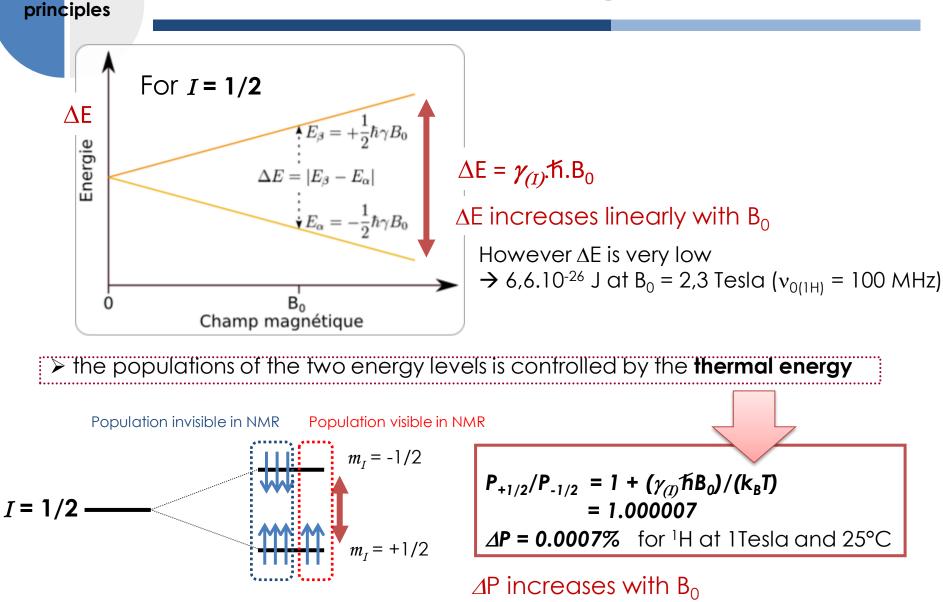


 ΔE does not depend on *I* !

Nevertheless NMR is nucleus selective because of the gyromagnetic ratio !

4. Sensitivity and Magnetic field

NMR



So the sensitivity increases with B₀ !!!

5. Sensitivity and Macroscopic magnetization NMR principles Needs of intense B₀ magnetic fields to increase the sensitivity Superconducting magnet $v_0 = \gamma B_0 / 2\pi$ Helium ports Helium tower $\gamma(^{1}H) = 26,7519.10^{7} \text{ rad } T^{-1} \text{ s}^{-1}$ Nitrogen ports Nitrogen tower $B_0(T)$ $v_0(^1H)$ (MHz) Insert sample here Metal plug 300 7 Vacuum Chamber 14 600 He ~ 5 m RUREN 21 900 1000 He 23.5 1000 Magnet Bo Earth's magnetic field = 5.10^{-6} T Insert Probe here At the right center of the coil B_0 is

At the right center of the coil B_0 is static, vertical and homogeneous

5. Sensitivity

Sensitivity : value relative to the proton considering 100% natural abundance.

$$S = (\gamma_X / \gamma_{1H})^3 \frac{(I_X + 1)I_X}{(I_{1H} + 1)I_{1H}}$$

Example
¹H: I=1/2,
$$\gamma = 26,7519.10^7 \text{ rad } \text{T}^{-1} \text{ s}^{-1} \rightarrow \text{S} = 1$$

³H: I=1/2, $\gamma = 28.535.10^7 \text{ rad } \text{T}^{-1} \text{ s}^{-1} \rightarrow \text{S} = 1.21$
¹⁹F: I=1/2, $\gamma = 25.181.10^7 \text{ rad } \text{T}^{-1} \text{ s}^{-1} \rightarrow \text{S} = 0.83$
³¹P: I=1/2, $\gamma = 10,841.10^7 \text{ rad } \text{T}^{-1} \text{ s}^{-1} \rightarrow \text{S} = 0.066$
¹³C: I=1/2, $\gamma = 6,7283.10^7 \text{ rad } \text{T}^{-1} \text{ s}^{-1} \rightarrow \text{S} = 0.016$
²⁹Si: I=1/2, $\gamma = -5.3188.10^7 \text{ rad } \text{T}^{-1} \text{ s}^{-1} \rightarrow \text{S} = 7,86.10^{-3}$

The proton is the most NMR sensitive nucleus

Sensitivity decreases quickly as soon as γ decreases because of $(1/\gamma)^3$

NMR principles

5. Receptivity

$$D = (\gamma_X / \gamma_{1H})^3 \frac{(ab.nat.)_X (I_X + 1) I_X}{(ab.nat.)_{1H} (I_{1H} + 1) I_{1H}}$$

Takes into account the natural abundance !

<u>spins 1/2</u> ¹H : 100% NA → D = 1

 $^{3}H: 0\% \text{ NA} \rightarrow D = 0$

 $^{19}\text{F}: 100\% \text{ NA} \rightarrow \text{D} = 0.834$

 $^{31}P:100\% \text{ NA} \rightarrow D=0,0665$

 $^{13}C:1,1\%$ NA \rightarrow D = 1,76.10⁻⁴

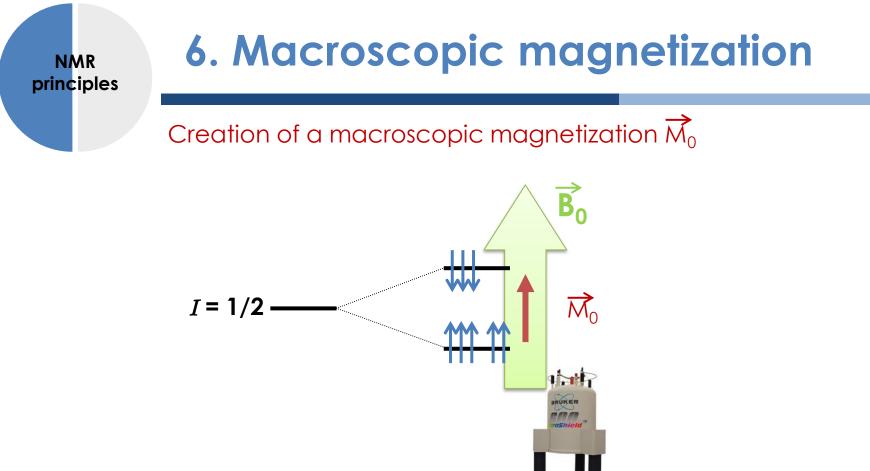
²⁹Si: 4,7% NA \rightarrow D = 3,69.10⁻⁴ ⁵⁷Fe: 2,2% NA γ = 0,87.10⁷ rad T⁻¹ \rightarrow D = 7,43.10⁻⁷ ... **Difficult...**

Quadrupolar spins $^{27}AI: I = 5/2; 100\% \text{ NA} \rightarrow D = 0.207$

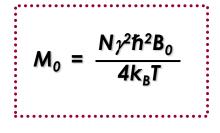
 ^{17}O : I = 5/2; 0,037% NA \rightarrow D = 1,08.10⁻⁵

 $^{43}Ca: I = 7/2; 0,145\% \text{ NA} \rightarrow D = 8,67.10^{-6}$

Difficult...



For a spin I=1/2



N: number of spins

Problem : M_0 is hidden in the static magnetic field B_0 (~10⁻⁶ of B_0)

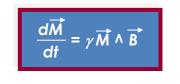
Impossible to measure directly !!!

How to measure it ???



Interaction of two magnetic moments; What happens to the magnetization $\overline{M_0}$ in the presence of a static magnetic field $\overline{B_0}$?

From the theorem of angular momentum, we know that there is a **time dependence** as follow : For two ordinary vectors



$$\vec{U} \begin{pmatrix} Ux \\ Uy \\ Uz \end{pmatrix} \quad \vec{V} \begin{pmatrix} Vx \\ Vy \\ Vz \end{pmatrix}$$

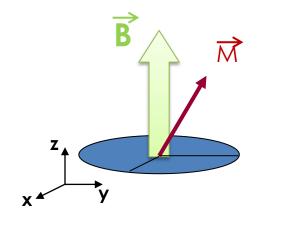
<u>dMy</u> dt

 $\underline{dM}z_{=0}$ dt

 $-\gamma B_0.Mx$

$$\vec{U} \wedge \vec{V} \begin{pmatrix} UyVz - UzVy \\ UzVx - UxVz \\ UxVy - UyVx \end{pmatrix}$$

 \vec{B} static and collinear to the z axis :



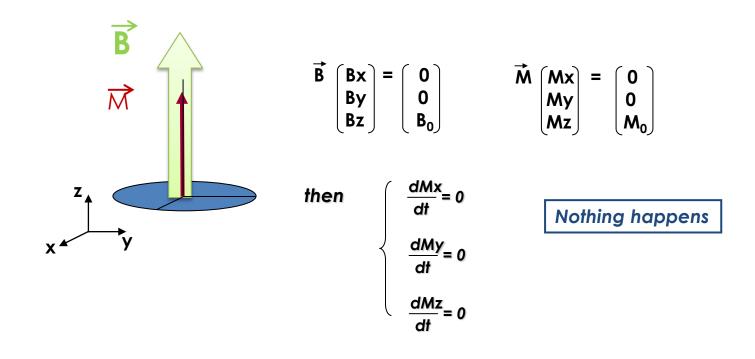
$$\vec{M} \begin{bmatrix} Mx \\ My \\ Mz \end{bmatrix} \qquad \vec{B} \begin{bmatrix} Bx \\ By \\ Bz \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ B_0 \end{bmatrix}$$
$$\frac{dMx}{dt} = \gamma B_0 \cdot My$$

Movement equations

Only the x and y components have a time dependence



At the thermodynamic equilibrium, B and M are collinear to the z axis



7. Concept of precession

Out of the thermodynamic equilibrium : \overrightarrow{M} is tilted from \overrightarrow{B} by an angle α

Then the initial conditions are

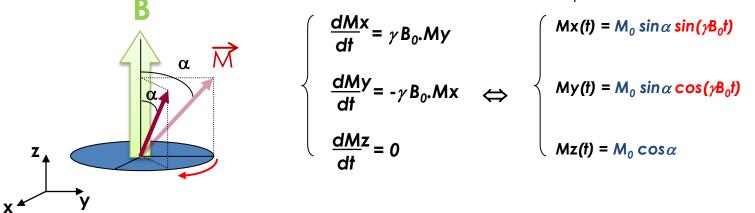
NMR principles

$$Mx(0) = 0$$

$$My(0) = M_0 \sin \alpha$$

$$Mz(0) = M_0 \cos \alpha$$

If we solve the differential equations



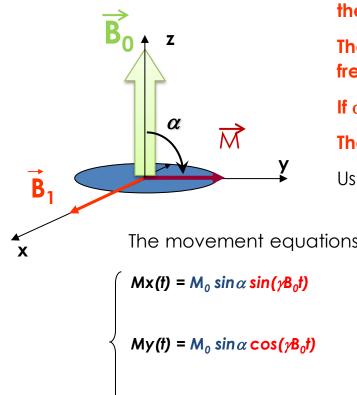
The movement equations describe a rotation of \overline{M} in the (xy) plane at the **angular rate** $\omega_0 = \gamma B_0$ The associated angular frequency is then $v_0 = \gamma B_0/2\pi$ i.e. the Larmor frequency !

NMR principles

7. Concept of precession

Goal of the NMR experiment : put \overrightarrow{M} out of the equilibrium

Best efficiency if $\alpha = 90^{\circ}$



If we apply a strong **B**₁ radio-frequency field perpendicular to B_0 along the x axis then we induce a rotation of M around the x axis

The frequency of the B₁ field must be $v_1 = v_0$ (Larmor frequency)

If α = 90° it is called a "90° pulse" or " π pulse"

The time to reach the (xy) plane is called $t_{90^{\circ}}$

Usually, t_{90° = from 1 to 10 µs

The movement equations are becoming :

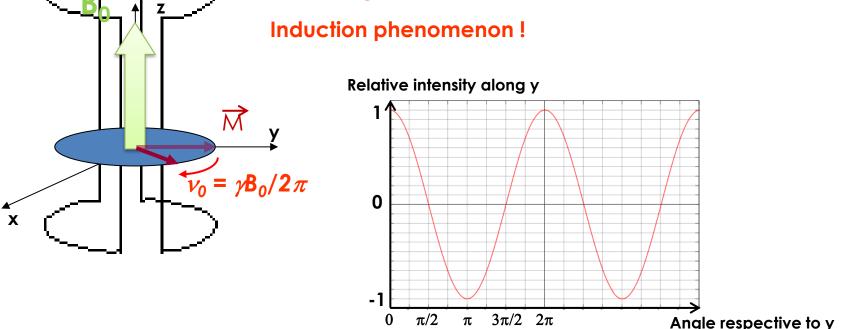
```
Mx(t) = M_0 sin(\gamma B_0 t)
                                                                    My(t) = M_0 \cos(\gamma B_0 t)
                                                     \Leftrightarrow
                                                                     Mz(t) = 0
Mz(t) = M_0 \cos \alpha
```

7. Concept of precession

If we switch off $\overrightarrow{B_1}$, \overrightarrow{M} start rotating in the (xy) plane at the v_0 frequency (Larmor frequency).

The macroscopic magnetization M is now measurable !

The variation of M inside a coil gives rise to an oscillating electric field !



Beware : the coil generating B_1 is different from the one generating B_0

The coil generating B_1 is the same as the coil allowing the recording of the NMR signal.